

$$\frac{X \text{ mmol pdt}}{0.0328} = \frac{\frac{3.04}{2H}}{\frac{4}{4H}}$$

$$X = 0.0498 \text{ mmol} = 9.8 \text{ mg product}$$

$$\frac{(9.8 \text{ mg})}{10.1 \text{ mg}} * 100 = 97 \text{ wt\%}$$

